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Section \_\_\_\_\_ Date \_\_\_\_\_ 10. A student drew the following electron box ...

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## Chemistry Olympiad Exams - American Chemical Society

Chapter 6 Standardized Test Practice 1. The actual number of atoms of each element contained in a molecule is given by the molecule's: A. empirical formula. C. mass percent formula. B. molecular formula. D. percent composition. 2. What is the molar mass of the sodium sulfate  $\text{Na}_2\text{SO}_4$ ? F. 71.06 g/mol G. 94.05 g/mol H. 142.05 g/mol J. 215.13 g/mol 3.

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