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$m = \rho V = 885 \text{ kg/m}^3 \times 0.001 \text{ m}^3 = 0.885 \text{ kg}$ For the air we see in Figure 2.7 that density is about 1 kg/m^3 so we get $m = \rho V = 1 \text{ kg/m}^3 \times 0.001 \text{ m}^3 = 0.001 \text{ kg}$ A more accurate value from Table A.5 is $\rho = 1.17 \text{ kg/m}^3$ at 100 kPa, 25°C. Borgnakke, Sonntag - Solution of. Thermodynamics 7th Edition ...

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