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4.2.2 How much energy is released when bonds form
to produce 2 mol of H_2O and 1 mol of CO_2 ? (4)

4.2.3 Hence calculate the energy transferred (ΔH) for
the combustion of 1 mol CH_4 . (2) 4.2.4 Draw a
labelled energy profile for this reaction. Include the
labels: ENTHALPY CHANGE (ΔH) ACTIVATION ENERGY
(E a) (6)

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scientist, the scientific establishment, and the public. Through the life of Hoyle, this book chronicles the triumphs, jealousies, rewards, and feuds of a rapidly developing scientific field, in a narrative animated by a cast of colourful astronomers, keeping secrets, losing their tempers, and building their careers here on Earth while contemplating the nature of the stars.

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