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1. Reactant

Stability/Reactivity: The  
more stable the reactant,  
the less reactive it will  
be. In terms of rates, this  
means that the more stable  
the reactant, the slower it  
will react. (The concept

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will produce a. Oxidation of  
a ketone will produce.  
Reduction of a carboxylic  
acid will pro... Reduction of  
a ketone will produce a (n)  
Carboxylic acid. No  
reaction. Aldehyde (or full



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reduction would be a primary alcohol. Secondary alcohol.

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What is the IUPAC names for  
this compound? a) 1-tert-

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butyl-2-butanol b)

5,5-dimethyl-3-hexanol c)

2,2-dimethyl-4-hexanol d)

1,1,1-trimethyl-3-pentanol

4. What is the IUPAC name for this structure?

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Identify the class of

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compounds each of the following molecules belong to (15pts). ROH ROR O R C O H R C O O-H R C O O C O C R O O-R RCN R C O NR 2 C H N O  
2) For each of the amides and amines state whether they are primary, secondary or tertiary (4pts).

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