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the right to the correct
crystal type on the left
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(a) has mobile electrons
in the crystal c covalent
molecular crystal (b) is
hard, brittle, and
nonconducting a
metallic crystal (c)
typically has the lowest
melting point of the four

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SHORT ANSWER

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Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest

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melting point of the four

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Review In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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