

**Modern Chemistry Chapter 10 Answers**

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**Modern Chemistry Chapter 10 Review Answers States Of ...**

Modern Chemistry Chapter 10 States of Matter. Modern Chemistry Chapter 10 States of Matter Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

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Chemistry Chapter 10. diffusion. effusion. Graham's law of effusion. permeability, the spontaneous, uniform spreading and mixing of particles cau.... a process in which gas particles pass through a tiny opening i.... the gas law that states that the rate of effusion is inversely....

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CHAPTER 10 REVIEW States of Matter SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

**10 States of Matter - Ms. Agostine's Chemistry Page**

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