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the bond angle CCN are a. sp^2 , 180° .
b. sp , 180° . c. sp^2 , 120° . d. sp^3 ,
 109° . 2. Which of the following
statements about an sp hybridized
carbon is FALSE? a. It is divalent. b. It
forms bonds that are linear. c. It has
two p orbitals. d.

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