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the conditions at the beginning of combustion, and p_4 , ρ_4 , and T_4 denote conditions at the end of combustion. Since the volume is constant, and the mass of the gas is constant, then $p_4 = \rho_3 = 11.3 \text{ kg/m}^3$. Thus, from the equation of state, $p_4 = \rho_4 RT_4 =$

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