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Solutions to exercises 1B.1(b) The mean speed is  $\sqrt{8RT/v_{\text{mean}} M}$  The mean translational kinetic energy is  $\frac{3}{2} RT$   $\frac{3}{2} kT$   $\frac{3}{2} M$   $\frac{3}{2} v_{\text{mean}}^2$  The ratios of species 1 to species 2 at the same temperature are  $\frac{E_{k,1}}{E_{k,2}} = \frac{v_{\text{mean},1}^2}{v_{\text{mean},2}^2}$  (i)  $\frac{1}{2} m v_{\text{mean},H}^2$  and  $\frac{1}{2} m v_{\text{mean},Hg}^2$   $\frac{200.6}{4.003} = \frac{1}{2} m v_{\text{rms}}^2$   $\frac{E_{k,1}}{E_{k,2}} = \frac{7.079}{1}$  (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

Atkins' Physical Chemistry Instructors Solutions Manual ...

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