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Conversion problems are easy to get confused. One way to think of

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Answers  
them is which is larger – quarts or gallons? Gallons are larger, so if you are converting from quarts to gallons the number of gallons will be a smaller number. Keeping that in mind, you can do a ‘commonsense’ check on your answer.

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answer key~~

The quantity in the problem is multiplied by one or more “conversion factors,” in which the numerator is equal to the denominator. Since the

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~~Answers~~ and denominator of the conversion factor are equal, multiplying by the conversion factor is like multiplying by 1 and thus does not change the value of the original quantity.

~~Answers to Conversion Factor~~

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Answers are easy to get confused. One way to think of them is which is larger - quarts or gallons? Gallons are larger, so if you are converting from quarts to gallons the number of gallons will be a smaller number. Keeping that in mind, you can do a



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answer in your head. If there are 100 cm in every meter, then 3.55 m equals 355 cm. To solve the problem more formally with a conversion factor, we first write the quantity we are given, 3.55 m. Then we multiply this quantity by a conversion factor, which is

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Answers as multiplying it by 1.

~~2.6: Problem Solving and Unit  
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Remember to check your work to

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Answers  
make sure an answer makes sense.

## ~~Measurements and Conversions Chemistry Quiz~~

conversion factor. Make sure the units cancel and you get the units you need. Always write your units

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Answers. Practice as many of the following as you need - the answers are below. Given: Conversion factors in your book, do NOT Google any other conversation factors. You can use the ones given on the shapes table. 1. How many pounds does

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2.00 kg of cheese weigh? 2. How many mL are in 0.50 quarts? 3.

## ~~CHM 130 Conversion Practice Problems~~

You will receive your score and answers at the end. question 1 of 3. Convert 40 degrees C to F. 212



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In these lessons, we will look into two methods that can be used for metric conversion problems. The first method uses the metric table and can also be regarded as a shortcut method. The second

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Answers method uses the unit fraction method. The following diagrams show some metric conversions for lengths and weights.

~~Metric Math Problems (solutions, examples, videos)~~

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mole conversions - practice  
problems It is important to be  
able to convert between units of  
mass and volume measurement



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Answers.' The mole is a unit for counting atoms and molecules representing 602,000,000,000,000,000,000,000 particles. This is often written as  $6.02 \times 10^{23}$  particles and is known as 'Avogadro's number.' Using the ...

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