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mass or takes up
space (volume) three
phases of matter.
gases, liquids, solids.
gases. have an
indefinite shape and
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as the “ design of

chemical products

and processes to

reduce or eliminate

the use and

generation of

hazardous

substances. ” 1,2

This definition and

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the concept of Green
Chemistry were first
formulated at the
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system, steady-state
processes with
multiple reactions,
Equation 10.1 in

moles becomes for
component i Where: v_{ij}
 v_{ij} is the stoichiometric
coefficient of species i
in reaction j in the
minimal set. ξ_j is
the extent of reaction
for the j th reaction in
...

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Solution Chemical And Engineering Reactions principles First Year Chapter 1 ...

$a A + b B \xrightarrow{C} x X$ with a catalyst C. Rate = $k [A]^m [B]^n [C]^p$. The exponents m , n , and p

- are the reaction order
- can be 0, 1, 2 or fractions
- must be determined by experiment!

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Rate Laws. Rate = k
 $[A]^m[B]^n[C]^p$. • If
 $m = 1$, rxn. is 1st
order in A Rate = k
 $[A]^1$.

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Chemistry Chemistry:
Principles and
Reactions When tin
comes in contact with

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the oxygen in the air, tin(IV) oxide, SnO_2 , is formed. $\text{Sn (s)} + \text{O}_2 \text{ (g)} \rightarrow \text{SnO}_2 \text{ (s)}$ A piece of tin foil, $8.25 \text{ cm} \times 21.5 \text{ cm} \times 0.600 \text{ mm}$ ($d = 7.28 \text{ g / cm}^3$) is exposed to oxygen.

When tin comes in contact with the oxygen in the air, tin

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A waterbed filled with
water has the
dimensions $8.0 \text{ ft} \times 7$

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