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molecule. written below. to leave out, not include.  
constant. not changing, continues. Monoxide. an  
oxide ...

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(25) Mendeleev organized the chemical elements  
based on their a. symbols. b. properties. c. atomic  
numbers. d. charges. Properties. A horizontal row in  
the periodic table is called a(n) a. family. b. group. c.  
octet. d. period. Period. The periodic ...

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Which of the following statements about gases is not true? There are strong, attractive forces between gas molecules. 3 The volume occupied by a sample of gas is 5,780 cm<sup>3</sup> of space. What is this ...

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the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. Bohr. b. Moseley. d. Ramsay. \_\_\_\_ 2. Mendeleev left spaces in his periodic table and predicted several ...

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radiation (5.1) the rays and particles-alpha particles, beta particles, and gamma rays-that are emitted by radioactive material. electromagnetic radiation (5.1) A form of energy exhibiting wavelike behavior as it travels through space; can ...

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Significant figures and appropriate units are required.

/6 a) 1.0 g of NaCl contains \_\_\_\_\_ grams of sodium b)

The mass of  $2.4 \times 10^{25}$  atoms of sodium is: c) 5.0 g of  $\text{CaCO}_3$  contains \_\_\_\_\_ moles of ...

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Chapter 5 1. In a double bond \_\_\_\_\_ electrons are shared between two atoms. a. 2 b. 3 c. 4 d. 6 2. Which molecule has a triple bond? a. H<sub>2</sub> b. F<sub>2</sub> c. Cl<sub>2</sub> d. N<sub>2</sub> 3. Which central atom has eight electrons around it when it forms the covalent compounds? a. AlCl<sub>3</sub> b. phosphorus pentachloride c. sulfur ...

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