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1. D This element has five peaks, meaning a total of five subshells. The final peak, which would be located in the 3p subshell, is slightly higher than the 3s peak to the left of it. A full 3s peak has two electrons, therefore there must be at least three electrons

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in the 3p subshell. The element that best fits this is phosphorus.  
2.

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the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

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