

## Chapter Assessment Nuclear Chemistry

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Nuclear Chemistry Chapter Introduction

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Chapter Assessment Nuclear Chemistry Chapter 21 □ Nuclear Chemistry: Part 2 of 9 In this lecture I'll teach you more about nuclear chemistry. I'll show you how to determine and balance the following types of ... Nuclear Chemistry - Isotopes & Ionizing Radiation A lesson on interpreting symbols of isotopes and the five main types of ionizing radiation.

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Now we can use Einstein's equation to determine the energy change of the nuclear reaction:  $E = (0.0000659 \text{ kg})(3.00 \times 10^8 \text{ m/s})^2 = 5.93 \times 10^{12} \text{ J}$ . This is almost 6 trillion joules given off. Test Yourself. The nuclear equation for the fission of uranium-238 is as follows: The molar mass in grams of each species is given for each particle.

Nuclear Energy □ Introductory Chemistry □ 1st Canadian Edition

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