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equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$ complete
ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

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Answer: 32 g 9.2 (a) 1 mole
of H atoms = 1 g 1 mole of
H₂ molecules = 2 × 1 = 2 g.
(b) 1 mole of N atoms = 14 g

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