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Ch. 9 Part 1: Stoichiometry

9.2 Ideal Stoichiometric Calculations ~~Stoichiometry~~ |

~~Chemical reactions and stoichiometry~~ | ~~Chemistry~~ |

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1 Solved Exercise 1st year
chemistry chapter 1
numericals Stoichiometry
Chapter 9 Section 1
Stoichiometry~~

Chapter 9 Section 1 Intro to
Stoichiometry including use
of molar mass and BEMR
(Balanced Equation Mole
Ratio)

~~9.1 Introduction to
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Concepts: Terms in this set

(10) Stoichiometry is the

branch of chemistry that

deals with elements in

compounds and with reactants

and products in chemical

reactions, focusing on. mass

relationships . the number

significant figures in an

answer to ...

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Stoichiometry Section 1

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Stoichiometry Section 1

Answers of N 2 are mixed

with 12.0 mol of H 2

according to the following

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Section 1 Stoichiometry

Answers: $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$
mc06se cFMSr i-vi - Chapter
9 Stoichiometry Section 1
Answers The balancing
numbers are known as the
coefficients of the
reaction.

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Reaction stoichiometry uses molar relationships to determine the amounts of unknown reactants or products from the amounts of known reactants or products.

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The relationship between the relative quantities of substances taking part in a reaction or forming a compound. What is

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~~Answers~~
Stoichiometry? Composition deals with the mass relationships of elements in compound; reaction involves the mass relationships between reactants and products in a chemical reaction. What is the difference between reaction ...

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CHAPTER 9 REVIEW

Stoichiometry SECTION 3
PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield.

2. 6.0 mol of N_2 are mixed with 12.0 mol of H_2 according to the

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Section 1 Stoichiometry

Following equation: $N_2(g)$
 $3H_2(g) \dots$

~~me06se eFMsr i vi~~

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compounds and with reactants
and products in chemical
reactions ...

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Chapter 9 Review

Stoichiometry Section 2

Answers 9-1 Introduction to

Stoichiometry Composition

Stoichiometry - deals with

mass relationships of

elements in compounds

Reaction Stoichiometry -

Involves mass relationships

between reactants and

products in a chemical

reaction I. Page 4/15

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