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PLAY. taste sour,  
turns litmus red,  
reacts with active  
metals to form  
hydrogen gas, react  
with bases to form  
water and a salt.

Characteristics of  
acids. taste bitter,  
turns litmus blue,  
slippery, reacts with

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and a salt.

Characteristics of  
bases.

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Bases: Properties,  
Names and Formulas  
What is an acid? An  
acid is

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Chapter 7 Acids,  
Bases, and Solutions  
Calculating a  
Concentration To

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calculate the concentration of a solution, compare the amount of solute to the amount of solution and multiply by 100 percent. For example, if a solution contains 10 grams of solute dissolved in 100 grams of solution, then its concentration can be reported as 10 percent.



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Chapter 7 acids and  
bases 1. Chapter 7

Acid-Base Equilibria

2. Acid-Base Theories

- Arrhenius theory  
(Nobel Prize) An acid  
is any substance that  
ionizes (partially or  
completely) in water  
to give hydrogen ions,  
 $H^+$  (which associate  
with the solvent to

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give hydronium ions,  
 $\text{H}_3\text{O}^+$ )  $\text{HA} + \text{H}_2\text{O}$   
 $\text{H}_3\text{O}^+ + \text{A}^-$  • A base:  
any substance that  
ionizes in water to  
give hydroxyl ions  
 $\text{OH}^-$ .

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bases - SlideShare*  
Acid + Base ? Salt +  
Water with evolution  
of heat The  
preparation of

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Solutions Cross  
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Common salt (sodium chloride) involves the neutralization reaction between hydrochloric acid and sodium hydroxide (base). The acid and base reacts to produce sodium chloride (salt), water and heat. The reaction is represented by the following equation:

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a substance that can act as either an acid or base (ex. water).

An amino acid has a basic amine group and an acidic carboxylic acid group, so it is also amphoteric. Unlike water, amino acids can carry multiple

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charges depending on their environment. At pH 7, most amino acids have a protonated amine group and a deprotonated carboxylic acid group.

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Sprint session in  
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Alternative. What is formed by the reaction of a non-metal oxide with water? [A] Acid [B]

Base [C] Salt [D]

Metal; ... The p H of a salt solution obtained by the reaction

between a strong acid and a strong base is...

[A] is 7 [B] is less than 7 [C] is more than 7

[D] cannot be



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## Bases And *Chapter 7 : ACIDS, BASES AND SALTS*

Chapter 7: Acids,  
Bases, and Solutions.  
solution. solvent.  
solute. properties.  
uniform mixture that  
contains a solvent  
and at least one  
solute. part of a  
solution present in the  
largest amount. the

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Substance that is present in a solution in a smaller amount....  
A solution has the same \_\_\_\_\_ throughout.

*exam quiz chapter 7  
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Bases 7.1 The Nature  
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7.2 Acid Strength 7.3

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7.4 Calculating the pH of  
Strong Acid Solutions

7.5 Calculating the pH  
of weak Acid

Solutions 7.6 Bases

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Properties of Salts 7.9

Acid Solutions in

Which Water

Contributes to the  $H^+$  +  
Concentration 7.10

Strong Acid Solutions

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in Which Water  
Contributes to the  $H^+$   
Concentration 7.11  
Strategy for solving  
Acid-Base Problems:  
A Summary

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01-10), Topic:  
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The strength of an acid. (7) 6. One of the products when you add an acid to a base. (5) 9. Acid found in your stomach. (12) 12. The process of adding acids to

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bases. (14) 14. One of the products of adding an acid to a base. (4) 16. A solid substance composed of positive and negative ions. (7)  
Down. 1. A substance that releases hydrogen ions ...

*Acids, Bases, Salts  
Crossword*

Chapter 7. Use of  
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Acid, Base, and Salt  
Salt, in chemistry,  
substance produced  
by the reaction of an  
acid with a base. A  
salt consists of the  
positive ion (cation) of  
an acid and the  
negative ion (anion) of  
a base.

*Chapter 7. Use of  
Acid, Base, and Salt -  
Teaching BD  
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7.1 7.2 7.3 7.4 7.5  
7.6 7.7 7.8 7.9 The  
Nature ...

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## 7.2... Chapter 7 Acids

Acids . Bases (i)

These are sour to taste. These are bitter to taste. (ii) Acids turn

blue litmus red. Bases do not change the colour of blue litmus.

(iii) Acids do not change the colour of red litmus. Bases turn red litmus blue. (iv)

With china rose indicator, these give

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Bases And  
Solutions Cross  
Word Puzzle

dark pink colour.  
Bases give green  
colour with china rose  
indicator. (v)

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Bases and Salts -  
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...

Acid + Base ? Salt +  
Water [with evolution  
of heat] The  
preparation of  
common salt (sodium

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chloride) involves the neutralization reaction between hydrochloric acid and sodium hydroxide (base). The acid and base reacts to produce sodium chloride (salt), water and heat. The reaction is represented by the following equation:

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Questions Science

Chapter 5 Acids,  
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Questions Very Short  
Answer Type

Questions. Question  
1. What is the test for  
acids and bases using  
litmus paper? Answer:

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Acids turn blue litmus  
paper red while bases  
turn red litmus paper  
blue. Question 2.

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