

Chapter 6 Thermochemistry Test

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162 CHAPTER 6: THERMOCHEMISTRY To convert the answer to joules, we write: 101.3 J 0.18 L atm =? ? x = ? w ?18 J 6.17 An expansion implies an increase in volume, therefore w must be ?325 J (see the defining equation for pressure-volume work.)

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164 CHAPTER 6: THERMOCHEMISTRY Substituting into the above equation: ?E = 483.6 × 103 J ? (8.314 J/mol?K)(398 K)(+1 mol) ?E = 4.80 × 105 J = 4.80 × 102 kJ 6.28 We initially have 6 moles of gas (3 moles of chlorine and 3 moles of hydrogen). Since our product is 6

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CHAPTER 6 Thermochemistry © Houghton Mifflin Company. All rights reserved. 115 14. Two metals of equal mass with different heat capacities are subjected to the same amount of heat. Which undergoes the smallest change in temperature? a) The metal with the higher heat capacity. b) The metal with the lower heat capacity.

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Chapter 6 - Thermochemistry - Mrs. Duffey - FHN

Chapter 6 Thermochemistryflow is determined into or out of the surroundings. Because ?Euniv = 0 by the first law of thermodynamics, ?Esys = !?Esurr; what happens to the surroundings is the exact opposite of what happens to the system. chapter 6 test chemistry thermochemistry Flashcards and ... Learn chemistry test Page 6/24