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Matter and Change

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Fluorine and xenon

combine to form

two dif-ferent

compounds. In one

compound, 0.853 g

of fluorine

combines with

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11.472 g of xenon.

In the other compound, 0.624 g of fluorine combines with 2.16 g of xenon. Do these data support the law of multiple proportions? Show your work ...

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EXAMPLE PROBLEM

11-1 For more practice converting from representative particles to moles, go to Supplemental Practice Problems in Appendix A. a c t i c e ! Ointments containing zinc oxide provide protection from sunburn and are

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11 used to treat some  
skin diseases.

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Assessment 5. How  
is a mole similar to  
a dozen? 6.

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CHAPTER Practice  
Problems 11.1 The  
Many Forms of  
Energy pages  
285–292 page 287  
1. A skater with a  
mass of 52.0 kg  
moving at 2.5 m/s  
glides to a stop  
over a distance of

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24.0 m. How much work did the friction of the ice do to bring the skater to a stop?

Key

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### **Energy and Its Conservation**

The Mole

Supplemental Problems KEY. 1.

Identify and calculate the

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14  
number of  
representative  
particles in each of  
the following  
quantities. a. 2.15  
moles of gold. b.  
0.151 mole of  
nitrogen oxide . c.  
11.5 moles of  
potassium bromide  
. 2. Calculate the  
number of moles of  
the substance that  
contains the



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following number  
of representative  
particles.

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Calculating Molar

Mass Converting

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Map Density, Ions,  
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Problems Physics

Test Prep: Studying  
for the ... and

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Problems for each chapter, as well as the Additional

Problems that appear in Appendix B ... 11. a. 139 cm

2.3 cm 320 cm<sup>2</sup> or  
3.2 102 cm<sup>2</sup> b.

3.2145 km 4.23 km  
13.6 km<sup>2</sup> 12. a.

13.78 g 11.3 mL

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### Supplemental Problems

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4 Godwit

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14 Associates paid \$60,000 for a 20-seat skybox at Memorial Stadium for eight professional football games. Regular seats to these games range from \$70 to \$150 each. At one game, an employee of Godwit entertained 18 clients. Godwit

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furnished food and  
beverages for the  
event at a cost of  
\$950.

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SOLUTIONS

MANUAL Section

11.1 Defining

Stoichiometry

pages 368–372

Practice Problems

pages 371–372 1.

Interpret the

following balanced

chemical equa-

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tions in terms of particles, moles, and mass.

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Problems 8.

Determine the molar mass of each of the 9. following

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14 compounds. a.

formic acid

(CH<sub>2</sub>O<sub>2</sub>) b.

ammonium

dichromate (NH<sub>4</sub>)<sub>2</sub>

Cr<sub>2</sub>O<sub>7</sub><sup>2-</sup> 42 27 -zsa .

What is the mass in

grams of each of

the following

quantities ? 3 a.

2.53 moles (Pb(NO<sub>3</sub>)<sub>2</sub>)

32 b. 4.62 moles

of magnesium

bromide (MgBr<sub>2</sub>)

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11 Calculate the  
number of moles in  
each of the 10. 11.

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challenging the  
brain to think  
improved and  
faster can be  
undergone by  
some ways.

Experiencing,  
listening to the

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11 new experience,  
adventuring,  
studying, training,  
and more practical  
comings and  
goings may back  
you to improve.

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of the following  
properties from an  
income tax

standpoint will be  
the best one for  
Jaun to contribute

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14 to his favorite  
charity? a. Stock  
acquired in 1992,  
basis = \$13,000,  
FMV = \$10,000.

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