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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

Assessment Chapter Test B - Ed W. Clark High School

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Assessment Chapter Test B - Wag & Paws

Solutions Manual Chemistry: Matter and Change • Chapter 12 239 CHAPTER 12 SOLUTIONS MANUAL Section 12.3 Liquids and Solids pages 415 – 424 Section 12.3 Assessment page 424 18. Contrast the arrangement of particles in solids and liquids. The particles are closer together in solids than in liquids because of intermolecular attractions.

States of Matter

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Her mother states she has been teased and bullied at school and she suspects that she tried to commit suicide by taking her pain pills. The patient is unresponsive to painful stimuli, with agonal respirations, and has vomited. She has a weak carotid pulse. Your first action is to: A. suction the patient's airway. B.

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Assessment: The Rise of Sumerian City-States

Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_\_ 13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. \_\_\_\_\_ 14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

Assessment Chapter Test A - Ed W. Clark High School

Holt Physics 1 Chapter Tests Assessment Chapter Test A Teacher Notes and Answers Forces and the Laws of Motion CHAPTER TEST A (GENERAL) 1. c 2. d 3. d 4. c 5. c 6. c 7. c 8. b 9. d 10. d 11. c 12. a 13. d 14. d 15. b 16. d 17. c 18. d 19. Forces exerted by the object do not change its motion. 20. An object at rest remains at rest and an

Assessment Chapter Test A

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