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Solution: $p_{abs} = p_{atm} + p_{gage} = 101.3 \text{ kPa} + 20 \text{ in. H}_2\text{O} \cdot 1248.8 \text{ Pa} / 1.0 \text{ in. H}_2\text{O} = 110.26 \text{ kPa}$ Ideal gas law: $\rho = p / RT = 110.26 \text{ kPa} / (1 \text{ kgK} \cdot 2077 \text{ J} / 1.293 \cdot 2 \text{ K} \cdot 1000 \text{ Pa} / 1 \text{ kPa}) = 0.181 \text{ kg/m}^3$ b.)

Situation: A sphere of 93 mm diameter contains an ideal gas. $T = 20 \text{ C} = 293.2 \text{ K}$ Find: Calculate the density of argon at a vacuum pressure of 8.8 psi. Properties:

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76 Solutions Manual • Fluid Mechanics, Fifth Edition 2.11 In Fig. P2.11, sensor A reads 1.5 kPa (gage). All fluids are at 20° C .

Determine the elevations Z in meters of the liquid levels in the open piezometer tubes B and C. Solution: (B) Let piezometer tube B be an arbitrary distance H above the gasoline-

Chapter 2 Pressure Distribution in a Fluid

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figure, the flat (turbulent) velocities do not resemble the parabolic laminar-flow profile of Prob. 3.3. (The discontinuity at $r = 1.75$ cm is an artifact—we need more data for $1.75 < r < 2.0$ cm.) The volume flow, $Q = \int u(2r)dr$, can be estimated by a

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