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2007 AP® CHEMISTRY FREE-RESPONSE QUESTIONS (Form B) © 2007 The College Board. All rights reserved. Visit apcentral.collegeboard.com (for AP professionals) and www.collegeboard.com/apstudents (for students and parents). GO ON TO THE NEXT PAGE. -10- (b) Chlorine gas, an oxidizing agent, is bubbled into a solution of potassium bromide.

2007 AP CHEMISTRY FREE-RESPONSE QUESTIONS (Form B)
The following comments on the 2007 free-response questions for AP®Chemistry were written by the Chief Reader, Eleanor Siebert of Mount St. Mary's College in Los Angeles, California. They give an overview of each free-response question and of how students performed on the question, including typical student errors.

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(a) Calculate the value of the standard free energy change, DG 298, for the reaction. (b) Determine the temperature at which the equilibrium constant, K eq, for the reaction is equal to 1.00. (Assume that 'Hqand 'Sq are independent of temperature.) (c) Calculate the standard enthalpy change, 'Hq, that occurs when a 0.256 mol sample of NF 3

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